





# Material Safety Data Sheet Boron MSDS

## **Section 1: Chemical Product and Company Identification**

Product Name: Boron

Catalog Codes: SLB1825

CAS#: 7440-42-8

**RTECS:** ED7350000

TSCA: TSCA 8(b) inventory: Boron

CI#: Not available.

Synonym:

Chemical Name: Boron

Chemical Formula: B

**Contact Information:** 

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US Sales: 1-800-901-7247

International Sales: 1-281-441-4400

Order Online: ScienceLab.com

CHEMTREC (24HR Emergency Telephone), call:

1-800-424-9300

International CHEMTREC, call: 1-703-527-3887

For non-emergency assistance, call: 1-281-441-4400

# **Section 2: Composition and Information on Ingredients**

## Composition:

Name	CAS#	% by Weight
Boron	7440-42-8	100

Toxicological Data on Ingredients: Boron: ORAL (LD50): Acute: 650 mg/kg [Rat]. 560 mg/kg [Mouse].

#### Section 3: Hazards Identification

**Potential Acute Health Effects:** Slightly hazardous in case of skin contact (irritant, permeator), of eye contact (irritant), of ingestion, of inhalation.

## **Potential Chronic Health Effects:**

Slightly hazardous in case of skin contact (permeator), of ingestion, of inhalation. CARCINOGENIC EFFECTS: Not available. MUTAGENIC EFFECTS: Not available. DEVELOPMENTAL TOXICITY: Not available.

#### **Section 4: First Aid Measures**

#### **Eve Contact:**

Check for and remove any contact lenses. In case of contact, immediately flush eyes with plenty of water for at least 15 minutes. Get medical attention if irritation occurs.

Skin Contact: Wash with soap and water. Cover the irritated skin with an emollient. Get medical attention if irritation develops.

Serious Skin Contact: Not available.

#### Inhalation:

If inhaled, remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get medical attention.

#### Serious Inhalation:

Evacuate the victim to a safe area as soon as possible. Loosen tight clothing such as a collar, tie, belt or waistband. Seek medical attention.

#### Ingestion:

Do NOT induce vomiting unless directed to do so by medical personnel. Never give anything by mouth to an unconscious person. If large quantities of this material are swallowed, call a physician immediately. Loosen tight clothing such as a collar, tie, belt or waistband.

Serious Ingestion: Not available.

## **Section 5: Fire and Explosion Data**

Flammability of the Product: Flammable.

Auto-Ignition Temperature: 580°C (1076°F)

Flash Points: Not available.

Flammable Limits: Not available.

Products of Combustion: Not available.

#### Fire Hazards in Presence of Various Substances:

Slightly flammable to flammable in presence of open flames and sparks, of heat. Non-flammable in presence of shocks.

## **Explosion Hazards in Presence of Various Substances:**

Slightly explosive in presence of open flames and sparks. Non-explosive in presence of shocks.

#### Fire Fighting Media and Instructions:

Flammable solid. SMALL FIRE: Use DRY chemical powder. LARGE FIRE: Use water spray or fog. Cool containing vessels with water jet in order to prevent pressure build-up, autoignition or explosion.

## **Special Remarks on Fire Hazards:**

Material in powder form, capable of creating a dust explosion. When amorphous boron is heated in dry ammonia, reaction proceeds with incandescence and hydrogen is evolved. Boron ignites in bromine vapor at 700 C. Boron ignites in chlorine @ 410 C. Fluorine attacks boron @ ordinary temperature and the resulting mass becomes incandescent. lodic acid attacks boron below 40 deg. C and the resulting mass becomes incandescent. Concentrated nitric acid and boron react so violently that the mass is raised to incandescence. A mixture of boron and sulfur becomes incandescent at 600 C. A reaction between nitrosyl fluoride and boron is accompanied by incandescence.

## **Special Remarks on Explosion Hazards:**

Fine dust dispersed in air in sufficient concentrations, and in the presences of an ignition source is a potential dust explosion hazard. Silver fluoride reacts explosively Boron at ordinary temperatures.

## **Section 6: Accidental Release Measures**

Small Spill: Use appropriate tools to put the spilled solid in a convenient waste disposal container.

## Large Spill:

Flammable solid. Stop leak if without risk. Do not touch spilled material. Use water spray curtain to divert vapor drift. Prevent entry into sewers, basements or confined areas; dike if needed. Eliminate all ignition sources. Call for assistance on disposal.

# **Section 7: Handling and Storage**

#### Precautions:

Keep away from heat. Keep away from sources of ignition. Ground all equipment containing material. Do not ingest. Do not breathe dust. Wear suitable protective clothing. If ingested, seek medical advice immediately and show the container or the label. Keep away from incompatibles such as oxidizing agents, acids.

#### Storage:

Keep container in a cool, well-ventilated area. Keep container tightly closed and sealed until ready for use. Avoid all possible sources of ignition (spark or flame).

## **Section 8: Exposure Controls/Personal Protection**

#### **Engineering Controls:**

Use process enclosures, local exhaust ventilation, or other engineering controls to keep airborne levels below recommended exposure limits. If user operations generate dust, fume or mist, use ventilation to keep exposure to airborne contaminants below the exposure limit.

**Personal Protection:** Safety glasses. Lab coat. Dust respirator. Be sure to use an approved/certified respirator or equivalent. Gloves.

#### Personal Protection in Case of a Large Spill:

Splash goggles. Full suit. Dust respirator. Boots. Gloves. A self contained breathing apparatus should be used to avoid inhalation of the product. Suggested protective clothing might not be sufficient; consult a specialist BEFORE handling this product.

**Exposure Limits:** Not available.

## **Section 9: Physical and Chemical Properties**

Physical state and appearance: Solid. (Amorphous solid powder or lumps.)

Odor: Not available.

Taste: Not available.

Molecular Weight: 10.81 g/mole

Color: Brown.

pH (1% soln/water): Not applicable.
Boiling Point: 2550°C (4622°F)
Melting Point: 2300°C (4172°F)
Critical Temperature: Not available.

Specific Gravity: 2.37 (Water = 1)
Vapor Pressure: Not applicable.
Vapor Density: Not available.

Volatility: Not available.

Odor Threshold: Not available.

Water/Oil Dist. Coeff.: Not available.
Ionicity (in Water): Not available.
Dispersion Properties: Not available.

Solubility:

Insoluble in cold water, hot water, diethyl ether. If finely divided, it is soluble in boiling sulfuric acid and most molten metals such as copper, iron, magnesium, aluminum, and calcium. Insoluble in alcohol.

## **Section 10: Stability and Reactivity Data**

Stability: The product is stable.

**Instability Temperature:** Not available.

Conditions of Instability: Heat, ignition sources, incompatible materials

Incompatibility with various substances: Reactive with oxidizing agents, acids.

Corrosivity: Non-corrosive in presence of glass.

#### Special Remarks on Reactivity:

Interaction of powdered boron and steam may become violent at red heat. The highly exothermic reactions with water might become combustive or explosive processes at sufficiently high temperatures and pressures. Incompatible with sodium peroxide, sodium carbonate, potassium nitrate, ammonia, iodic acid, nitric acid, nitrosyl fluoride, phosphorous, silicon, silver fluoride, lead dioxide, cesium carbide, rubidium carbide, cupric oxide, nitric oxide.

Special Remarks on Corrosivity: Not available.

Polymerization: Will not occur.

## **Section 11: Toxicological Information**

Routes of Entry: Inhalation. Ingestion.

Toxicity to Animals: Acute oral toxicity (LD50): 560 mg/kg [Mouse].

Chronic Effects on Humans: Not available.

Other Toxic Effects on Humans: Slightly hazardous in case of skin contact (irritant, permeator), of ingestion, of inhalation.

Special Remarks on Toxicity to Animals: Not available.

Special Remarks on Chronic Effects on Humans: Not available.

#### **Special Remarks on other Toxic Effects on Humans:**

Potential Health Effects: Skin: May cause skin irritation. Eyes: May cause eye irritation. Inhalation: May cause respiratory tract and mucous membrane irritation. Prolonged or repeated inhalation may affect respiration, and metabolism (weight loss, anorexia) Ingestion: Prolonged or repeated ingestion may affect the brain, liver, heart. Chronic poisoning (from ingestion, skin absorption, or absorption from body cavities or mucous membranes) causes anorexia, weight loss, vomiting, mild diarrhea, skin rash, alopecia, convulsions (or other nervous system disturbances), and anemia

## **Section 12: Ecological Information**

Ecotoxicity: Not available.

BOD5 and COD: Not available.

#### **Products of Biodegradation:**

Possibly hazardous short term degradation products are not likely. However, long term degradation products may arise.

**Toxicity of the Products of Biodegradation:** The products of degradation are less toxic than the product itself.

Special Remarks on the Products of Biodegradation: Not available.

## **Section 13: Disposal Considerations**

#### Waste Disposal:

Waste must be disposed of in accordance with federal, state and local environmental control regulations.

## **Section 14: Transport Information**

DOT Classification: CLASS 4.1: Flammable solid.

Identification: : Flammable solid, n.o.s UNNA: 1325 PG: III

Special Provisions for Transport: Not available.

## **Section 15: Other Regulatory Information**

Federal and State Regulations: TSCA 8(b) inventory: Boron

Other Regulations:

OSHA: Hazardous by definition of Hazard Communication Standard (29 CFR 1910.1200). EINECS: This product is on the

European Inventory of Existing Commercial Chemical Substances.

Other Classifications:

WHMIS (Canada): Not controlled under WHMIS (Canada).

DSCL (EEC):

R11- Highly flammable. R22- Harmful if swallowed. S16- Keep away from sources of ignition - No smoking.

HMIS (U.S.A.):

Health Hazard: 1

Fire Hazard: 1

Reactivity: 0

**Personal Protection: E** 

National Fire Protection Association (U.S.A.):

Health: 1

Flammability: 1

Reactivity: 0

Specific hazard:

**Protective Equipment:** 

Gloves. Lab coat. Dust respirator. Be sure to use an approved/certified respirator or equivalent. Safety glasses.

#### **Section 16: Other Information**

References: Not available.

Other Special Considerations: Not available.

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